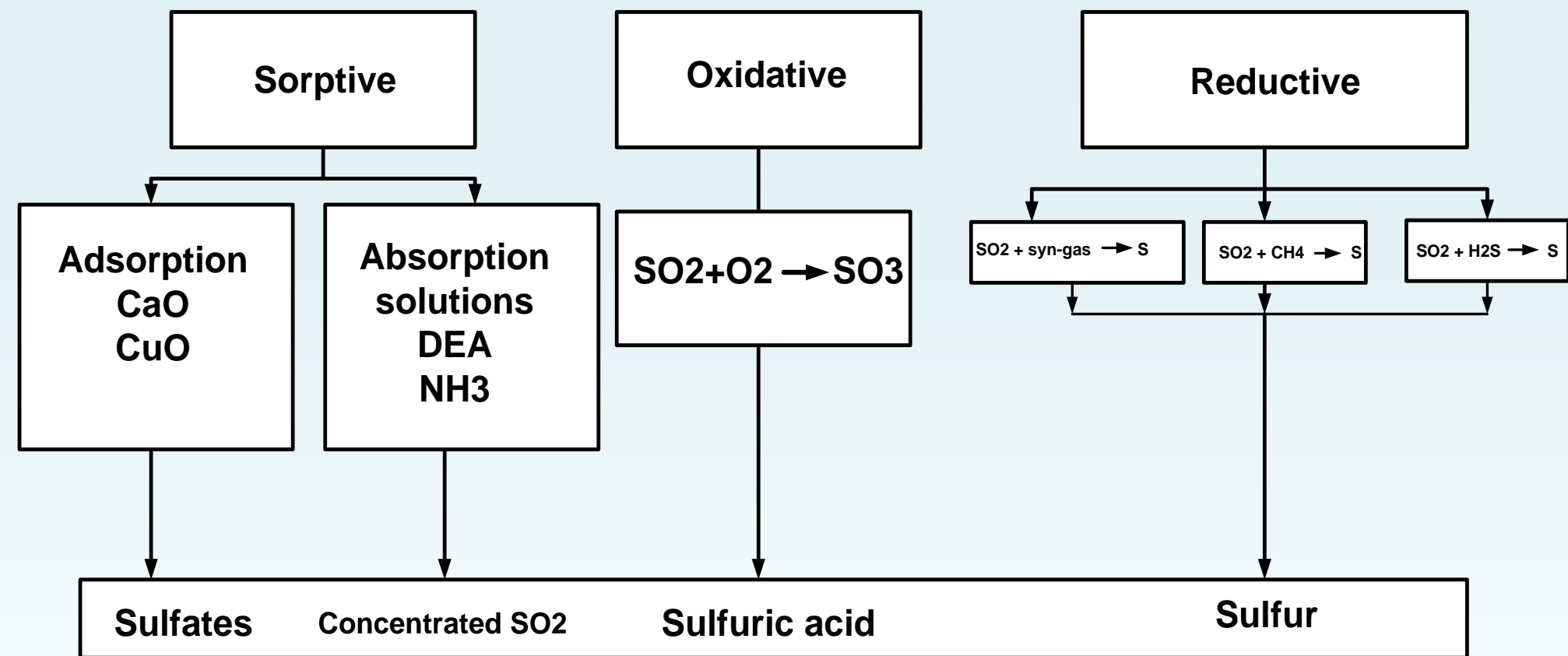




New catalysts and advanced technology for catalytic reduction of sulfur dioxide from emissions of nonferrous smelters

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Modern methods of SO₂ abatement



Catalytic methods leading to formation of elementary sulfur are the most prospective ways for substantial reduction of sulfur dioxide emissions. Sulfur is non-toxic product, no problems with transportation and storage. High demand of world market in sulfur. Current price is over 600 \$ per metric tone

The requirements to the catalysts to be developed:
High temperature reduction of SO₂ with methane
 $SO_2 + CH_4 \leftrightarrow S (H_2S) + H_2O + CO_2$
 High thermal stability (the range of operating temperatures is 700-960°C)
 Resistance to coking

Low temperature reduction of SO₂ with syn-gas
 $SO_2 + H_2(CO) \leftrightarrow S + H_2O(CO_2)$
 High selectivity to sulfur
 Suppression of formation of side products carbon disulfide, carbon sulfide oxide.



The enterprises of non-ferrous metallurgy are the main factor of environment pollution with sulfur dioxide

SO₂ is resulting from the roasting of non-ferrous ores

Typical emissions:
 ca. 1 000 000 tons of SO₂ per year (Polar Division of Open Joint Stock Company Mining and Metallurgical Company Norilsk Nickel - Russia)
 ca. 500 000 tons of SO₂ per year (Balkhash Mining and Metallurgical Company – Republic of Kazakhstan)



Laboratory set-up at BIC

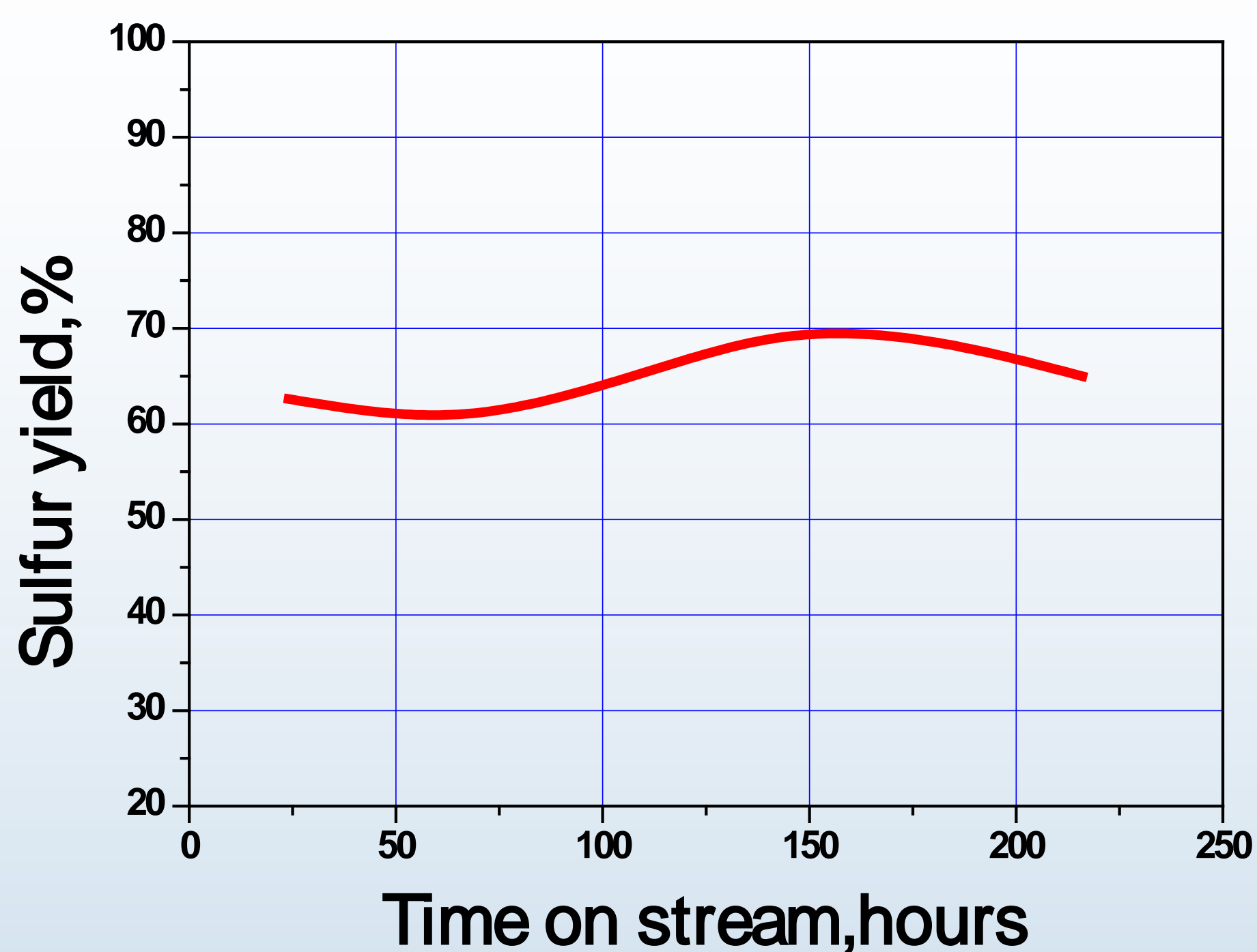


Testing the catalysts performance in BIC pilot plant

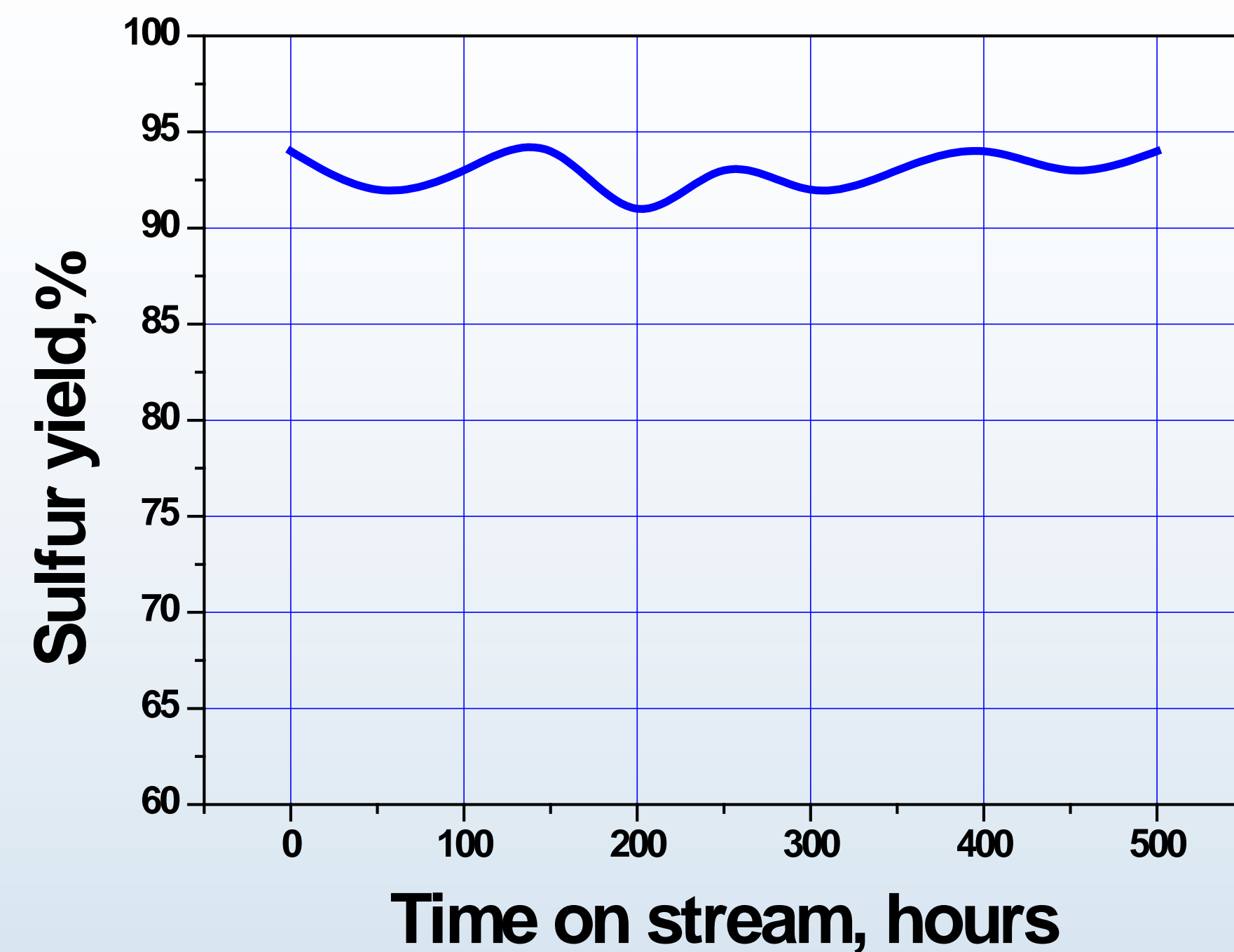


Pilot plant for testing the catalysts under realistic conditions

MAIN RESULTS OF PILOT TESTS



High temperature reduction of SO₂ with methane
 Temperature: 910 ÷ 965°C,
 supported oxide catalyst,
 GHSV = 1250 ÷ 1320 h⁻¹ ■



Low temperature reduction of SO₂ with syn-gas
 Temperature: 400 ÷ 500°C,
 supported oxide catalyst,
 GHSV = 1450 h⁻¹.

Comparative analysis of process versions

PROCESS CONDITIONS	SO ₂ REDUCTION WITH METHANE	SO ₂ REDUCTION WITH SYN-GAS
Temperature, °C	910 ÷ 960	400 ÷ 600
Sulfur yield, %	60-65	> 80

to high prospects of low-temperature sulfur dioxide reduction with syn-gas and its substantial advantages over